## **LESSON PLAN**

DISCIPLINE:	SEMESTER:	NAME OF THE TEACHING FACULTY:
MATH AND SCIENCE	FIRST	Miss. Dipti Laxmi Bhuyan, Lecturer in Chemistry, Govt. Polytechnic, Koraput

SUBJECT: ENGG. CHEMISTRY	NO. OF. DAYS PER WEEK CLASS ALLOTED	From – 9/11/2020 to 31/3/2021 THEORY PRACTICAL		
WEEK	CLASS DAY			
	1 <sup>ST</sup>	-Introduction, Matter and its states. -Atomic structure: fundamental	Introduction to chemistry lab, about safety measures, about maintenance of practical records. Introduction to the students about use of different lab	
	-	particles (electron, proton and neutron), their properties.	equipments and how to handle them safely.	
1 <sup>st</sup>	3 <sup>RD</sup>	<ul> <li>Atomic number and mass no.</li> <li>Rutherford's atomic model and its drawback.</li> </ul>		
	4 <sup>TH</sup>	<ul> <li>Drawback of Rutherfords atomic model.</li> <li>Bohr's Atomic model</li> </ul>		

⊃ <sup>nd</sup>	1 <sup>ST</sup>	<ul> <li>Drawbacks of Bohr's atomic model.</li> <li>Bohr-Bury scheme</li> <li>Quantum Number, types</li> </ul>	Exp. 1, preparation and study of properties of $CO_2$ gas, explanation of theory with equations.
	2 <sup>ND</sup>	-Detail of quantum number. -Aufbau's principle, Hund's rule, Electronic configuration	Checking of rough practical record and demonstratation of the experiment.
	3 <sup>RD</sup>	-Doubt clearing class of Atomic structure. -Question practice.	
	4 <sup>TH</sup>	-Introduction to Chemical bonding(definition, cause and types	
3 <sup>rd</sup>	1 <sup>ST</sup>	)-Ionic Bonding and Covalent bonding(Definition and formation of compounds with examples)	Expt. Conducted by the students.
	2 <sup>ND</sup>	<ul> <li>-Coordinate bonding(Definition and formation of compounds with examples)</li> <li>- discussion and Doubt clearing of Chemical bonding</li> </ul>	Correction of practical records, discussion of viva questions of the expt.
	3 <sup>RD</sup>	-Basics of acid and bases. -Arrhenius concept of acid and	

		base.	
	4 <sup>TH</sup>	-Draw backs of Arrhenius theory and Bronsted-Lowry theory with examples.	
	1 <sup>st</sup>	<ul><li>-Conjugate Acid-Base pair with examples.</li><li>-Drawbacks of Bronsted-Lowry Theory</li></ul>	Exp. 2. Preparation and study of properties of ammonia gas. Explanation Of Theory With Equations.
<b>⊿</b> th	2 <sup>ND</sup>	-Lewis Theory of acid and base with examples.	Checking of rough practical record and demonstratation of the experiment.
4	3 <sup>RD</sup>	<ul> <li>-Neutralization Reaction with examples.</li> <li>-Doubt clearing of acid base concept</li> </ul>	
	4 <sup>TH</sup>	-Definition of salt and types of salt.	
	1 <sup>st</sup>	<ul> <li>Definitions of atomic weight, molecular weight,</li> <li>Equivalent weight.</li> <li>Determination of equivalent weight of Acid, Base and Salt.</li> </ul>	Expt. Conducted by the Students.
5 <sup>th</sup>	2 <sup>ND</sup>	- Molarity , Normality -Related Numericals	Checking of practical records and discussion of viva questions of expt. 2.
	3 <sup>RD</sup>	-Molality and related numericals.	
	<b>4</b> <sup>TH</sup>	-pH of solution and numericals	
	1 <sup>ST</sup>	- Importance of pH in industry.	Exp. 3. Crystalization of CuSO <sub>4</sub> . Explanation Of Theory With

<b>c</b> th		- Doubt clearing.	Equations.
6	2 <sup>ND</sup>	-Electrochemistry: Definition	Checking of rough practical record and demonstratation of
		and types (Strong & weak) of	the experiment.
		Electrolytes with example.	
	3 <sup>RD</sup>	- Process of Electrolysis, its	
		Mechanism with different	
		example.	
	4 <sup>TH</sup>	- Faraday's 1st and 2nd law of	
		Electrolysis.	
	1 <sup>ST</sup>	Numericale	Funt Conducted by the Students
	1	-Numericals,	Expt. Conducted by the Students.
		-Industrial application of	
	2 <sup>ND</sup>	Electrolysis- Electroplating.	Checking of practical records and discussion of viva quastions
	2	<ul><li>-Corrosion and its types.</li><li>Definition of Corrosion, Types</li></ul>	Checking of practical records and discussion of viva questions of expt. 3.
		of Corrosion	of expt. 3.
		- Atmospheric Corrosion,	
<b>7</b> <sup>th</sup>		Waterline corrosion.	
	3 <sup>RD</sup>	- Mechanism of rusting Iron	
		only.	
		- Protection from Corrosion by	
		Alloying and Galvanization	
	4 <sup>TH</sup>	Basics of Organic chemistry	
		-Types of organic compound on	
		the basis carbon skeleton.	
	1 <sup>ST</sup>	-Hydrocarbons:	Exp. 4. Acid Base Titration. Explanation Of Theory With

		definitions, general formula,	Equations.
		examples.	
<b>الد</b>		-Rules for IUPAC system of	
<b>Q</b> <sup>th</sup>		nomenclature.	
0	2 <sup>ND</sup>	-Some more Rules for IUPAC	Checking of rough practical record and demonstratation of
		system of nomenclature.	the experiment.
	3 <sup>RD</sup>	Huckles rule, Aromatic	
		compounds.	
		-Practice of IUPAC nomenclature	
	<b>T</b> 11		
	4 <sup>TH</sup>	-Doubt clearing of organic	
		chemistry.	
	CT.	-revision	
	1 <sup>ST</sup>	-Definition of Mineral, ores,	Expt. Conducted by the Students Acidimetry.
		gangue with example.	
		-introduction to the extraction	
	ND	of minerals	
+6	2 <sup>ND</sup>	-Ore Dressing	
<b>Q</b> <sup>III</sup>		-Gravity separation, magnetic	Expt. Conducted by the Students Alkalimetry.
<u> </u>	PD	separation,	
	3 <sup>RD</sup>	-Froth floatation & leaching	
	тц	-Calcinations,.	
	4 <sup>TH</sup>	-Roasting.	
		-Smelting & examples of flux,	
	ст	slag	
	1 <sup>ST</sup>	-Electro refining, & Distillation	Checking of practical records and discussion of viva questions
			of expt. 4.

th	2 <sup>ND</sup>	-Definition of alloy. Types of alloys with example. -amalgam	Exp. 5. Test of acid radicals. Discussion regarding Basic ideas of acid and basic radicals, aim and basic steps of the test.
10 <sup>th</sup>	3 <sup>RD</sup>	<ul> <li>-Composition and uses of Brass,</li> <li>Bronze, Alnico, Duralumin</li> <li>-Revision of Inorganic</li> <li>Chemistry.</li> </ul>	
	4 <sup>TH</sup>	- Water Treatment : Sources of water, Soft water, Hard water, hardness, types of Hardness.	
th	1 <sup>ST</sup>	<ul><li>-Removal of hardness by lime soda method</li><li>- Advantages of Hot lime over cold lime process.</li></ul>	Checking of rough practical record and demonstratation of the experiment.
11"	2 <sup>ND</sup> 3 <sup>RD</sup>	-Organic Ion exchange method -Question discussion and	Expt. Conducted by the Students.
	<b>T</b> 11	Revision.	
	4 <sup>TH</sup>	<ul> <li>Definition of lubricant, Types.</li> <li>Uses of Graphite, Oils, Grease.</li> </ul>	
	1 <sup>ST</sup>	- Purpose of lubrication, Revision.	Checking of practical records and discussion of viva questions of expt. 5.
	2 <sup>ND</sup>	- Definition and classification of	
		fuel	
		-Definition of calorific value of fuel	
17 <sup>th</sup>		- Choice of good fuel.	
	3 <sup>RD</sup>	- Composition and uses of	

		diesel, petrol, kerosene.	
		Producer gas and Water gas	
	4 <sup>TH</sup>	- composition & uses of LPG,	
		CNG and coal gas.	
		- revision of chapter fuel.	
	1 <sup>ST</sup>	-Basic ideas about polymer	Exp. 6. Test of basic radicals (known).
		-Definition of Monomer,	
		Polymer, Homo-polymer, Co-	
		polymer with example.	
	2 <sup>ND</sup>	-Degree of polymerization	Checking of rough practical record and demonstratation of
		-Difference between	the experiment.
		Thermosetting and	
		Thermoplastic,	
1 - th		-Composition and uses of	
13		Polythene,	
	3 <sup>RD</sup>	-Poly-Vinyl Chloride and	
		Bakelite.	
		-,	
	4 <sup>TH</sup>	-Natural Rubber	
		- Definition of Elastomer	
	1 <sup>ST</sup>	Vulcanisation of Rubber.	Expt. Conducted by the Students.
		-Advantages of Vulcanised	
		rubber over raw rubber.	
h a th	2 <sup>ND</sup>	-Question discussion and	Test of unknown acid and basic radicals.
⊥4		Revision of polymer	
	3 <sup>RD</sup>	-introduction regarding modern	

	agriculture. -Chemicals in Agriculture:	
4 <sup>TH</sup>		
4	-Pesticides: Insecticides,	
	herbicides, fungicides with	
	Examples and uses	
1 <sup>ST</sup>	-Bio Fertilizers: Definition,	Test of unknown salt.
	examples and uses.	
	- Question discussion and	
	Revision of polymer	
 2 <sup>ND</sup>	-Doubt clearing and Question	Checking of practical records and viva voice.
	discussion	
3 <sup>RD</sup>	-Previous year Questions and	
	probable questions Discussion.	
4 <sup>TH</sup>	-Previous year Questions and	
	probable questions Discussion.	